

Mech'cheM 2025

New forces in Mechanochemistry

Mechanochemical Polymer Synthesis: Where We Are Now

Jeung Gon Kim^{a,b}

^a Department of Chemistry, Jeonbuk National University, Jeonju, South Korea

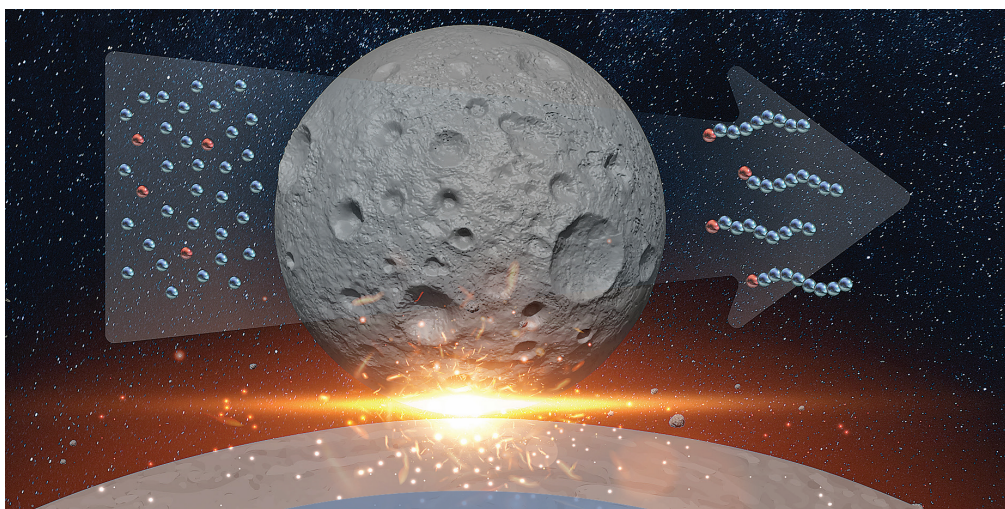
^b Research Institute for Materials and Energy Science Institution, Jeonju, South Korea

jeunggonkim@jbnu.ac.kr

Over the last decade, we have witnessed remarkable advancements in mechanochemistry through numerous successful reports. More importantly, we now have a deeper understanding of the processes occurring during mechanochemical reactions. In this presentation, I aim to highlight the ability of mechanochemistry to thread monomers and produce unique polymers that are inaccessible through conventional solution-based synthesis.¹

Examples from our group and others have demonstrated how mechanochemical ball-milling can effectively control molecular weight, dispersity, and polymer composition.² Notably, the solid-state nature of mechanochemistry eliminates issues with insoluble monomer combinations.³ Additionally, polymer degradation back to monomers and insights into mixing dynamics during polymerization have been observed.⁴

Through this presentation, the audience will appreciate that mechanochemical synthesis is not only an exciting area of research but also extends beyond its appeal as a "green" technology.



¹ A. Krusenbaum, S. Grätz, G. T. Tigineh, L. Borchardt, J. G. Kim, *Chem. Soc. Rev.* **2022**, *51*, 2873-2905.

² G. S. Lee, H. W. Lee, H. S. Lee, T. Do, J.-L. Do, J. Lim, G. I. Peterson, T. Frišćić, J. G. Kim, *Chem. Sci.* **2022**, *13*, 11496-11505.

³ G. S. Lee, H. S. Lee, N. Kim, H. G. Shin, Y. H. Hwang, S. J. Lee, J. G. Kim, *Macromolecules* **2024**, *57*, 9408-9418.

⁴ H. W. Lee, K. Yoo, L. Borchardt, J. G. Kim, *Green Chem.* **2024**, *26*, 2087-2093

